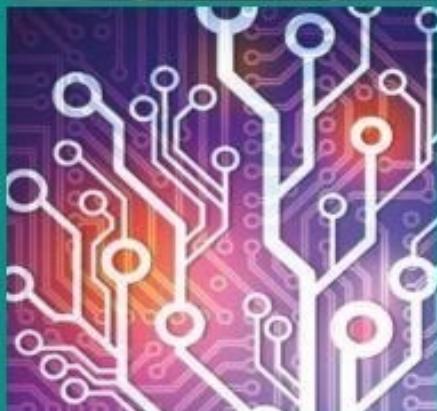
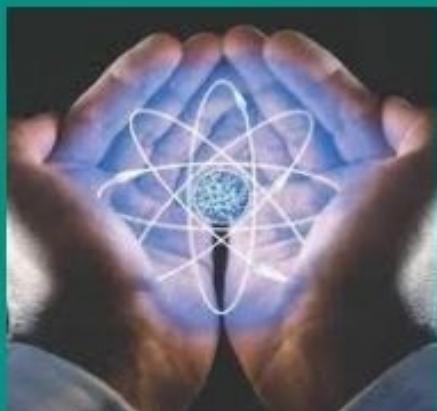


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# Academia Open



*By Universitas Muhammadiyah Sidoarjo*

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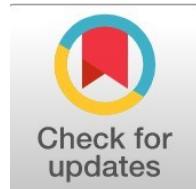
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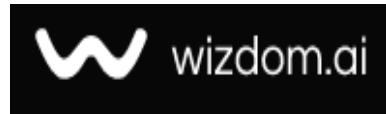
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# **Assessment of Trace Elements in Water, Sediments, and Fish Tissues from the Al-Gharraf River**

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## **Abstract**

**General Background:** Effective **General Background:** Trace elements are persistent environmental contaminants that accumulate in aquatic systems and pose ecological and health risks.

**Specific Background:** The Al-Gharraf River in southern Iraq receives inputs from agricultural, industrial, and domestic activities, increasing concern about metal enrichment in water, sediments, and fish.

**Knowledge Gap:** Despite regional studies, data on the simultaneous distribution and bioaccumulation of cadmium and zinc in multiple environmental compartments of this river remain limited. **Aims:** This study assessed concentrations of cadmium and zinc in water, sediments, and fish tissues across three months to determine contamination levels and ecological implications. **Results:**

Findings showed pronounced elevation of both metals in December compared with November and January, with the highest accumulation recorded in sediments and in fish gills and liver, while muscles exhibited comparatively lower concentrations. The consistent month-to-month variation demonstrated active metal inputs and dynamic environmental interactions that facilitate bioaccumulation. **Novelty:** This study provides integrated, compartment-based evidence of trace-metal distribution in a key river system using synchronized sampling of water, sediment, and multiple fish tissues. **Implications:** The results underscore the urgent need for strengthened monitoring and management strategies to mitigate metal pollution and safeguard ecosystem integrity and public health.

## **Highlight :**

- Shows increased cadmium and zinc levels in December across all sample types.
- Fish organs, especially gills and liver, display clear bioaccumulation patterns.
- The findings underline the need for stricter environmental monitoring.

**Keywords :** Trace Metals, Al-Gharraf River, Water, Sediments, Fish Tissues

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## Introduction

In the latest years there has been rising interest regarding metal contaminations in the aquatic environments, seemingly due to the increasing toxicity and remarkable persistence within aquatic systems. The aquatic environment consists mainly of three sources of minerals: water, sediments, and living organisms [1-3]. Several studies have been conducted on the pollution of rivers with trace elements, and the overcontamination of these elements has become a concern for the health of the aquatic ecosystem [4,5]. Trace elements are highly stable, persistent for long periods, and are largely unaffected by other environmental factors due to their ability to form multiple complex compounds. Therefore, it is difficult to remove them [6,7]. Thus, this pollution causes great environmental stress to aquatic organisms [8]. Moreover, its examination in rivers, lakes, benthic organisms, and sediments is vital to monitor toxicity and non-biodegradability, which has been the subject of several research works [9-11]. Bioaccumulation of trace elements can lead to toxic effects on tissues and organs. In fact, trace elements disrupt cellular events, including growth, proliferation, differentiation, damage-repairing processes, and apoptosis [12]. Although the toxic effects of trace elements depend on the form and the way of exposure. Interruptions of intracellular homeostasis include damage to lipids, proteins, enzymes, and DNA via the production of free radicals [13]. Sediments act as a reservoir for trace elements, which can subsequently accumulate in bottom-feeding fish, making these organisms valuable bioindicators of metal pollution. These fish can be potentially useful bioindicators for tracking metal pollution in aquatic environments. Furthermore, trace elements in water and sediment after fertilization reveal a complicated interaction between biological and abiotic variables in the system [14, 15]. In fact, some trace elements in the sediment are endowed with physicochemical properties that allow them to be mobilized and released back into the water column via hydrodynamics, biogeochemical processes, and anthropogenic activities [16, 17]. Furthermore, the bioaccumulation of trace elements in organisms can be passive or selective [18]. This study aims to assess trace element levels in water, sediments, and fish tissues in the Gharraf River, Shastra District, to determine the extent of contamination and its potential environmental impact.

## Materials and Methods

The study was conducted on the Al-Gharraf River in Al-Shastra District, Dhi Qar Governorate, southern Iraq. Samples were collected monthly for three months at the study station, from November 2024 to January 2025. The study included three types of samples "water, sediment and fish". Water samples were collected from the middle of the river using plastic (polyethylene) 5 L bottles with three replicates. A few drops of HNO<sub>3</sub> were added as a stabilizing agent to stabilize the elements in the water. For the sediment, the samples were collected from the middle of the river using a grab sampler of Van Veen. They were kept in nylon bags and stored in boxes according to the IMRP. Fish samples were collected from the study area using gill nets (25 mm mesh) and then placed in an insulated cork container containing ice until they reached the laboratory. The trace elements were extracted from the water after being digested using the method of APHA (19). The trace elements in the sediments were extracted using the method of Yi et al. (20). For fish samples, R.O.P.M.E. (21) method was used to measure trace elements. After preparing the samples, they are analyzed with atomic absorption spectrophotometer device.

## Statistical Analysis

SPSS version 23 was used for the statistical analysis, and the mean  $\pm$  standard deviation (mean  $\pm$  SD) was the outcome. The study groups were compared one-way ANOVA.

## Results and Discussion

Table 1 shows the concentrations of cadmium and lead in water. The results showed that the lowest cadmium concentration was (0.07 mg/L) in January, while the highest concentration was (0.23 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ . also the same table showed that the lowest zinc concentration was (1.61 mg/L) in January, while the highest concentration was (3.27 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ .

Table 1: "Concentration of trace elements in water during the study months".

Months \ Metals	Zinc (mg/L) Mean $\pm$ SD	Cadmium(mg/L) Mean $\pm$ SD
November	2.53 $\pm$ 0.34 <sup>b</sup>	0.15 $\pm$ 0.003 <sup>b</sup>
December	3.27 $\pm$ 0.21 <sup>a</sup>	0.23 $\pm$ 0.006 <sup>a</sup>
January	1.61 $\pm$ 0.18 <sup>c</sup>	0.07 $\pm$ 0.003 <sup>c</sup>
LSD	0.13	0.02

- Note:** Each value represents mean  $\pm$  S.D values with non-identical superscript (a, b or c...etc), were considered significantly differences ( $P \leq 0.05$ ).
- SD:** Standard deviation.
- LSD:** Least Significant Difference

Table 2 shows the concentrations of cadmium and zinc in the sediments. The results showed that the lowest average concentration of cadmium was (0.24 mg/L) in January, while the highest average concentration was (0.41 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ . Table 2 also shows the concentration of zinc in the sediments. The results showed that the lowest average concentration of zinc was (2.04 mg/L) in January, while the highest average concentration was (3.72 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ .

**Table 2:** "Concentration of trace elements in sediments during the study months".

Months Met.als	Zinc (mg/L) Mean ±SD	Cadmium(mg/L) Mean ±SD
November	2.89.±0.21 <sup>b</sup>	0.34.±0.05 <sup>b</sup>
December	3.72.±0.45 <sup>a</sup>	0.41 ±0.04 <sup>a</sup>
January	2.04.±0.18 <sup>c</sup>	0.24.±0.01 <sup>c</sup>
LSD	0.12	0.08

- Legend as in Table (1)

Table 3 shows the cadmium and zinc concentration in fish tissues. In the gills, the lowest cadmium concentration was (0.13 mg/L) in January, while the highest concentration was (0.27 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ . In liver, the lowest cadmium concentration was (0.18 mg/L) in January, while the highest concentration was (0.30 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ . In the muscles, the lowest cadmium concentration was (0.090 mg/L) in January, while the highest concentration was (0.25 mg/L) in December. Significant differences were found between the three months, at a probability level of  $p \leq 0.05$ . The same table shows zinc concentration in fish tissues. In the gills, the lowest zinc concentration was (1.95 mg/L) in January, while the highest concentration was (3.62 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ . In liver, the lowest zinc concentration was (1.95 mg/L) in January, while the highest concentration was (3.58 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ . In the muscles, the lowest zinc concentration was (1.73 mg/L) in January, while the highest concentration was (3.48 mg/L) in December. Significant differences were found between the three months at a probability level of  $p \leq 0.05$ .

**Table 3:** "Concentrations of zinc and cadmium in tissues (gills, kidneys and muscles) during the study months"

Months Site	Zinc (mg/L) Mean ±SD			Cadmium(mg/L) Mean ±SD		
	Muscles	Liver	Gills	Muscles	Liver	Gills
November	0.13±0.03 <sup>b</sup>	0.26.±0.04 <sup>b</sup>	0.16.±0.02 <sup>b</sup>	2.57.±0.41 <sup>b</sup>	2.74.±0.32 <sup>b</sup>	2.67.±0.25 <sup>b</sup>
December	0.25 ±0.01 <sup>a</sup>	0.30 ±0.07 <sup>a</sup>	0.27 ±0.10 <sup>a</sup>	3.48.±0.16 <sup>a</sup>	3.62.±0.53 <sup>a</sup>	3.58.±0.37 <sup>a</sup>
January	0.09.±0.001 <sup>c</sup>	0.18.±0.03 <sup>c</sup>	0.13.±0.005 <sup>c</sup>	1.73.±0.21 <sup>c</sup>	1.95.±0.22 <sup>c</sup>	2.01.±0.18 <sup>c</sup>
LSD	0.04	0.04	0.02	0.10	0.14	0.16

- Legend as in Table (1)

The study revealed elevated concentrations of cadmium and zinc in water, sediments, and fish tissues—including gills, kidneys, and muscles—throughout the sampling period. Aquatic ecosystems often act as final receptors for various contaminants, particularly trace metals that threaten environmental and human health (22). Although metals may originate from natural geological weathering, anthropogenic inputs—such as industrial discharge, agricultural activities, and inadequate waste disposal—generally represent the dominant sources of enrichment in aquatic environments (23). Additional contributions may arise from atmospheric deposition and uncontrolled waste release, both of which can increase metal loads (24,25). Because organisms in aquatic habitats remain in constant contact with their environment, the effects of metal pollution are more pronounced, leading to bioaccumulation and potential biomagnification through food webs (26,27). Fish are widely used as bioindicators due to their capacity to assimilate metals from both water and sediments (28,29). Given the rising dependence on fish as a nutritional resource, evaluating metal concentrations is essential to ensure food safety and assess possible health risks (30-32). Furthermore, sediments function as major sinks for trace metals and provide an important record of long-term contamination in aquatic systems (33-35)..

## Conclusion

The findings of the present study confirm that the aquatic environment is exposed to considerable inputs of cadmium and zinc, leading to their accumulation in sediments and fish tissues. These results highlight the urgent need for stricter environmental monitoring and improved management practices to reduce metal contamination and protect ecosystem and public health.

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